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Biosynthesis of poly-3-hydroxybutyrate by *Rhodococcus pyridinivorans* using unrelated carbon sources

Naima Khan*, Nazia Jamil

Abstract

Background: Polyhydroxyalkanoates are intracellular polymers comparable to synthetic plastic in their thermostable and elastomeric properties. PHAs are produced by bacteria under various nutrient – nitrogen – stress conditions.

Methods: Bacteria were isolated from hot water springs of Tatta Pani Kashmir and Karachi Mangrove forest Pakistan. Polyhydroxyalkanoate detection agar was used to isolate PHA producing bacteria and several carbon sources such as glucose, glycerol and palmitic acid were used for relative biomass and biopolymer productions. PHA was extracted by solvent extraction method using sodium hypochlorite and chloroform. Extracted polymer was characterized by Fourier transfer infrared spectroscopy (FTIR).

Results: *Rhodococcus pyridinivorans* NK19 (KY703220) produced up to 60% PHA with glucose, 40% with palmitic acid and 58% with glycerol as carbon sources. FTIR spectrum confirmed the polymer produced as poly -3, hydroxybutyrate. A peak at 1720 cm⁻¹ of FTIR confirmed the presence of PHB monomers in the polymer extracted.

Conclusion: *Rhodococcus pyridinivorans* NK19 produced short chain length PHA interchangeably known as P3HB while utilizing unrelated carbon sources up to 60%.



Introduction

Solid waste management has been a major problem since the industrial revolution in the 1950s. Incineration of plastics produces toxic waste, solid as well as gases, and results in incomplete degradation. Approximately 187 million tons of plastic waste are introduced into the environment annually [1]. Bioplastics can be used as alternatives of synthetic plastics. Polyhydroxyalkanoates (PHAs) are thermostable and elastomeric and degrade completely under favorable environmental conditions [2]. They are produced by bacteria as intracellular inclusion bodies. Bacteria accumulate PHAs and related compounds under various environmental stresses including limiting nutrients such as nitrogen thus making them better candidate to survive and withstand natural selection pressures. Polyhydroxyalkanoates are biodegradable, biocompatible polyesters with various industrial applications that almost parallel synthetic polyesters [3]. Most common PHAs are polyhydroxybutyrates (P3HB) and has wide number of applications due to its polystyrene like properties [4]. PHAs are cytoplasmic inclusions may range in size from 0.2 to 0.5 μm [5]. The molecular weight of PHB depends upon its source, growth situations and extraction method. Its molecular weight ranges from 50,000 to a million Da. Organically created, PHB is a semi crystalline isotactic stereo systematic polymer with 100% R configuration, which permits enhanced degradability [5].

Polyhydroxyalkanoates have wide range of applications in medical as well as industrial fields due to their living tissue compatibility, zero toxicity and biodegradability [6]. Their wide range of physical and chemical properties enable them to mimic extracellular matrix thus making in-vitro studies more reliable and natural. They have been used as sutures, adhesion barriers and valves to guide tissue repair and in regeneration devices such as cardiovascular patches, articular cartilage repair scaffolds, bone graft substitutes, and nerve guides [7,8].

Polyhydroxybutyrate is the PHA that has been most frequently studied and characterized to date. It is a natural and biodegradable aliphatic homopolymer that is composed of monomers of four carbon atoms. It is soluble in some organic solvents and insoluble in water. It has the chemical formula $(\text{C}_4\text{H}_6\text{O}_2)_n$, which corresponds to 55.81% carbon, 7.03% hydrogen, and 37.16% oxygen by weight. The industrial interest in the use of P(3HB) emerged during the 1960s when its thermoplastic properties were first described. Its synthesis was first conducted in 1971 via the polymerization of a racemic mixture of *b*-butyrolactone using a catalyst system of triethyl aluminum and water, which resulted in a stereo regular polymer that is partially optically inactive. In additional studies, P(3HB) was biotechnologically obtained from bacteria with a low molar mass (1×10^4 Da) and crystallinity (29%). P(3HB) can be processed as a classic thermoplastic as it can resist wide temperature range (from 30 °C to 120 °C). P(3HB) is non-toxic and its degradation produces 3-hydroxybutyric acid, a normal constituent of human blood. Thus, it can be used both in products that

assimilate with human or animal tissue and for human consumption [9]. In this study we isolated several bacterial strains with ability to produce PHAs under salt stress. Ability to grow optimally in presence of salt makes bacterial isolates more desirable [10]. Salt tolerant bacteria minimize the growth of non-tolerant bacteria thus reducing the cost of production as there will be no need to sterilize the salt containing growth media. We used glycerol, glucose, palmitic acid and propionic acid as carbon sources for relative biomass and polymer production.

Methods

Bacterial strain isolation and screening for salt tolerant PHA producers

Different bacterial strains were isolated from hot water springs of Kashmir, Pakistan and mangrove forest of Karachi, Pakistan. We collected rhizosphere soil of mangroves in sterile containers, similarly fine stones and water samples were also collected from hot water springs and surrounding area. Samples were serially diluted and bacterial colonies were obtained by spread plate method. Isolated bacterial strains were then subjected to screening based on salt tolerance and PHA production. PHA detection agar was supplemented with 0.5 M NaCl for screening purpose. Bacterial strains that were able to tolerate salt and produce PHA were identified using 16S ribosomal RNA gene. Deoxyribonucleic acid was extracted from bacteria and the conserved 16S rRNA gene was amplified and sequenced. Sequence homology was done using NCBI-BLAST to determine the similarity of isolated bacterial strain with already submitted sequences at the NCBI GenBank. For screening of PHA producing strains, Nile red and Nile blue A stains were used to stain the colonies. After 24 hours of incubation plates were checked for fluorescence. For further detection of PHA, Sudan Black B staining was done to visualize intracellular PHA granules.

Growth kinetics of isolated strain for pH, temperature and salt stress

Various pH (4, 6.8, 8, and 10), temperature (10°C, 25°C, 37°C and 45°C), nitrogen concentration (0.2, 0.5, 1 and 2%) and salt concentrations (0.1M, 0.5M, 1M, 2M, and 3 M) were used to optimize growth of isolated bacterial strains [11]. Bacterial growth was monitored at two-hour interval by measuring optical density of the broth culture at A_{660} nm. PHA production was also estimated at an interval of six hours. The experiment was conducted in shake flask culture and PHA was isolated using a modified method using [12] ultrasound assisted solvent extraction with chloroform and sodium hypochlorite.

Relative production of biomass and PHA using different carbon sources

Glucose – 2%, glycerol – 2% and palmitic acid – 0.5% was used as carbon sources for optimal PHA production. PHA detection agar (PDA) [13] was used to conduct shake flask experiment. All production experiments were conducted at optimal pH, temperature, and salt conditions. Growth and production of PHA was

monitored for a period of 72 hours. After every two-hours, optical density of the culture was checked at $A_{600\text{ nm}}$.

Characterization of extracted polymer

Fourier transfer infrared spectroscopy (FTIR) of the extracted polymer was done to check the presence of different PHA related functional groups. The samples were dissolved in chloroform and deposited on KBr disc and were scanned between 400 to 4000/cm wave number [14].

Results

Bacterial strain identification and growth optimization

Bacterial colonies of *Rhodococcus pyridinivorans* (KY703220) stained with Nile blue and Nile red showed bright pink fluorescence when viewed under UV as shown in the figure 1. P3HB producing bacteria give fluorescence when grown in the presence of Nile red [15]. Sudan Black staining revealed purple-black granules inside rod shaped bacteria (figure 2). Sudan Black has high affinity for carbonyl functional group containing molecules that therefore, it is assumed that it must have attached with P3HB granules. Evolutionary history of selected strain, *Rhodococcus pyridinivorans* (KY703220) was detected using molecular evolutionary genetic analysis software (MEGA 7.2). Neighbor joining tree method gave optimal tree with the sum of branch length = 1.05268421 as shown in figure 3. The analysis involved nine nucleotide sequences selected from NCBI database having maximum similarity with the isolated strain, *Rhodococcus pyridinivorans* (KY703220).

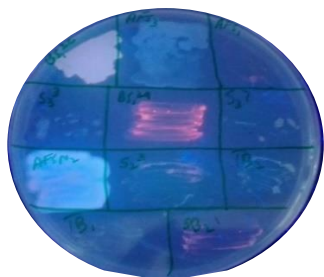


Figure 1: Isolated strain showing pink fluorescence when grown in PHB detection agar supplemented with 0.5 M NaCl and 0.5mg/ml Nile Red [13].

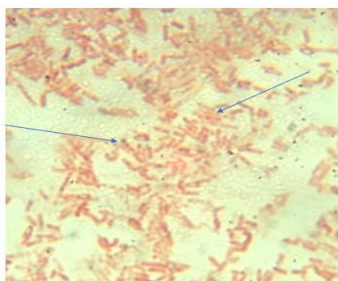


Figure 2: Intracellular P3HB granules visible when producer strain was stained with Sudan black dye and observed at 1000x magnification.

Relative growth of *Rhodococcus pyridinivorans* under abiotic stresses

In this study we isolated fifty bacterial strains from extreme environments of Pakistan. After screening, several isolates were found to be PHA producers. Thus, our focus was to isolate salt tolerant PHA producing strains. *Rhodococcus pyridinivorans* (KY703220) was able to produce P3HB in as much as 3M salt media (figure 4). The isolated strain was able to grow and produce P3HB at different NaCl concentrations. However, at concentrations above 1M its growth rate reduced. Salt tolerant bacteria are an excellent candidate for the production of PHA as they might reduce sterility cost and will be able to grow in extreme conditions [16].

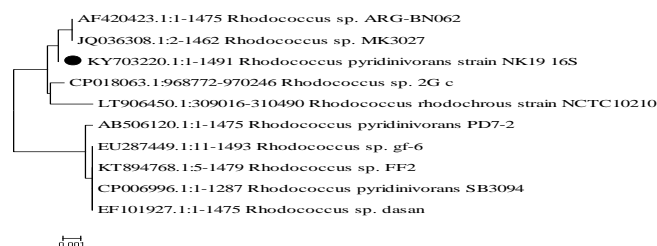


Figure 3: Phylogenetic analysis of *Rhodococcus pyridinivorans* NK19 by neighbor joining tree method by MEGA.

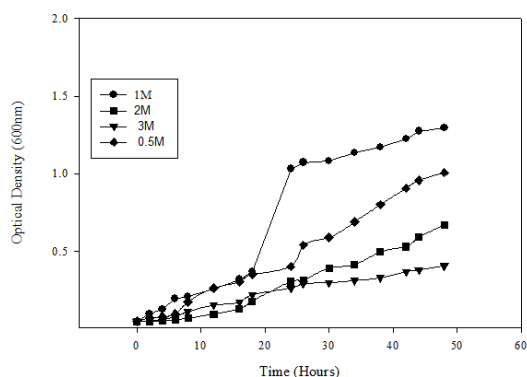


Figure 4: Growth optimization of salt tolerant *Rhodococcus pyridinivorans* (KY703220) at different salt concentrations.

PHA production

Glucose, Palmitic acid and Crude glycerol was used as carbon sources for the production of P3HB. Growth kinetics observed showed that strain was growing best with glucose giving optical density of 4.9, with glycerol OD recorded was 3.5 while with palmitic acid 1.435 after 72 hours of incubation at 37°C and 0.5 M salt (~ 3%) as shown in figure 5. *Rhodococcus pyridinivorans* (KY703220) showed different trends for carbon sources as shown in the figure 6. When glucose was used as carbon source *Rhodococcus pyridinivorans* (KY703220) was able to produce 60 % PHB of dry cell weight. While with other carbon sources – glycerol and palmitic acid – the production was relatively low 40% and 58% respectively (figure 6). In this study we isolated bacterial strain that was not only able to produce P3HB using different types of carbon sources but can also tolerate

high salt concentrations thus making it suitable for further large-scale studies. P3HB produced by *Rhodococcus pyridinivorans* (KY703220) can be used as it in various commercial application or modified by various metabolic as well as physical methods [17-19].

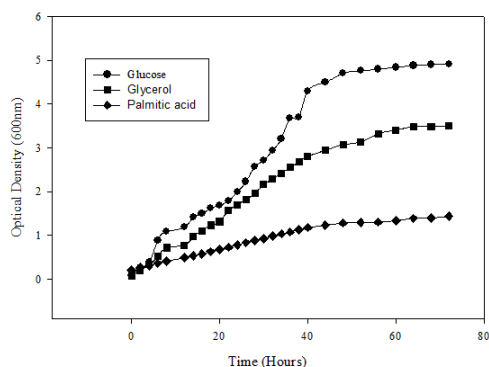


Figure 5: Growth kinetics of *Rhodococcus pyridinivorans* (KY703220) using various carbon sources at 0.5M NaCl concentration.

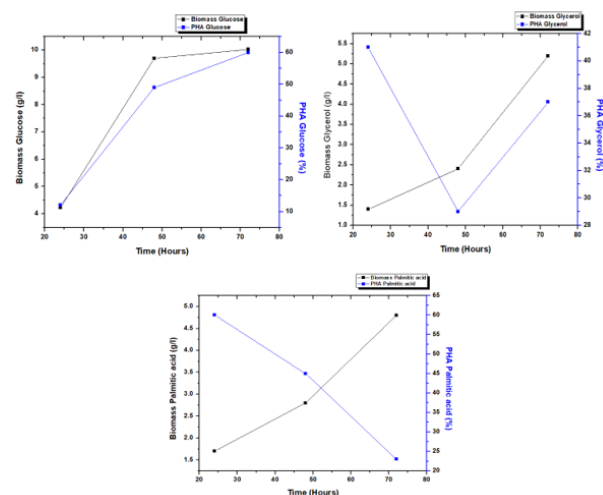


Figure 6: Comparative production of Biomass and P3HB using different carbon sources – glucose, glycerol and palmitic acid.

Characterization of extracted polymer

Fourier transfer infrared spectroscopy detected the presence of carbonyl functional group at 1720cm^{-1} . This functional group is particularly associated with short chain length PHA.

Discussion

Bio-plastic production has faced many setbacks due to the fact that its production is rather costly as compared to synthetic polymers. However, its importance cannot be denied and this problem can be fixed by discovering bacterial species that can survive under severe abiotic stress and utilize a number of unrelated carbon sources. *Rhodococcus pyridinivorans* (KY703220), although not a typical halophile, was able to tolerate NaCl concentrations as much as 1M (5.8%). Most other environmental isolates need vigorous aseptic conditions as other microbial contaminants can also compete with

them under same growth conditions. Halophiles do not require such conditions as other mesophile are not able to grow under high salt stresses. This is an advantage that can be used in biotechnology to reduce production cost of various products as important as bioplastics [22]. Cost effective production of PHB can also be ensured by using cheap carbon sources. Carbon sources of organic and inorganic nature originated as waste from various capacities either domestic or commercial are being investigated to produce PHB [23]. Production of PHA, as much as 0.14g/l , has been previously reported by *Rhodococcus opacus* B4 using glucose, acetate and hexadecane and *Rhodococcus equi* produced 38% poly-3(hydroxybutyrate) using crude palm kernel oil [16,17]. In this study, isolated bacterial strain was able to accumulate 40% PHB using palmitic acid and 58 % with glycerol.

FTIR spectrum of the extracted PHA showed a strong peak at 1720.39 cm^{-1} which was reported as bond stretching vibrations of carbonyl (C O), this region is supposedly a polyhydroxybutyrate marker peak and is associated with amorphous region of the biopolymer. This absorption region also indicate that the polymer has maximum crystallinity another attribute of the short chain length PHA [24, 25]. Thus, the polymer extracted from *Rhodococcus pyridinivorans* (KY703220) is confirmed to be PHB. Due to high crystallinity of the scl-PHA it has a limited number of applications, however the polymer can be modified by increasing content of 3-hydroxyvelarate or 4-hydroxybutyrate and can be used in various fields including biomedical and cosmetics industry.

Author Contributions

Naima Khan conducted experimental work and constructed the initial draft of the manuscript. Nazia Jamil designed experimental work as well as provided guidelines for the manuscript preparation.

Competing Interest

All authors declare no conflicts of interest in this paper.

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